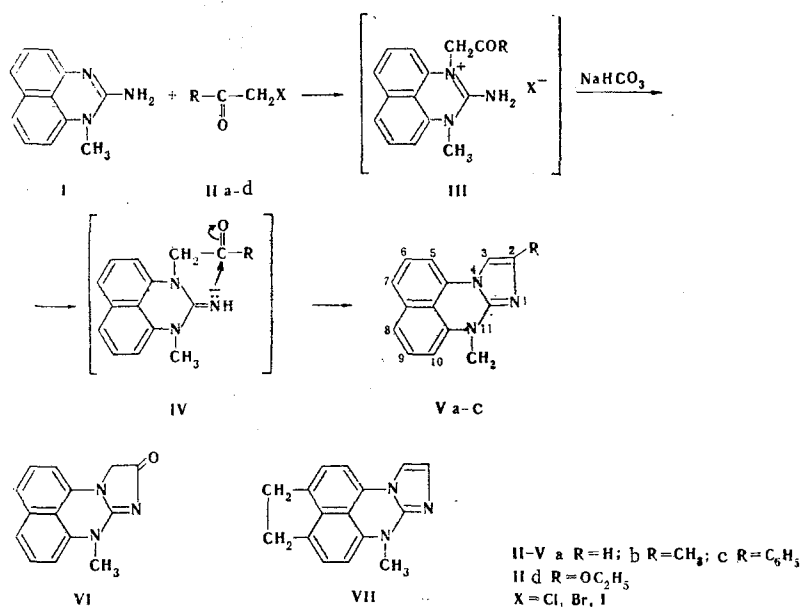


SYNTHESIS OF IMIDAZO[1,2-*a*]PERIMIDINE
DERIVATIVES

V. A. Anisimova, A. F. Pozharskii,
A. M. Simonov, and V. V. Dal'nikovskaya

UDC 547.856.7:547.785.5

Derivatives of a new heterocyclic system – imidazo[1,2-*a*]perimidine (V) – were obtained by reaction of 2-amino derivatives of perimidine and aceperimidine with α -halo ketones, α -halo aldehydes, and α -halo esters in the presence of sodium carbonate under very mild conditions (heating of aqueous alcohol solutions on a water bath for 1-3 h).



The reaction apparently proceeds through a step involving the formation of salt III and imine IV; however, these intermediates are distinguished by high instability and cannot be isolated even when the pro-

TABLE 1. Compounds of the Imidazo[1,2-*a*]perimidine Series

Com- pound	mp, °C (crystallization solvent)*	Empirical formula	Found, %			Calc., %			Yield, %
			C	H	N	C	H	N	
Va	159 (dec., alcohol)	C ₁₄ H ₁₁ N ₃	76,2	4,9	19,0	76,0	5,0	19,0	91
Vb	153—154 (benzene)	C ₁₅ H ₁₃ N ₃	76,6	5,7	18,1	76,6	5,6	17,8	73
Vc	141—142 (aqueous alcohol)	C ₂₀ H ₁₅ N ₃	80,7	5,3	14,1	80,8	5,1	14,1	92
VI	260 (dec., DMFA)	C ₁₄ H ₁₁ N ₃ O	70,7	4,5	18,0	70,9	4,7	17,7	96
VII	183 (dec., alcohol)	C ₁₆ H ₁₃ N ₃	77,6	5,5	16,9	77,7	5,3	17,0	90

* All of the compounds obtained, except oxo derivative VI, can also be purified by chromatography on aluminum oxide with elution by chloroform (R_f 0.9-0.93).

Rostov State University. Scientific-Research Institute of Physical and Organic Chemistry, Rostov-on-Don. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, No. 11, pp. 1578-1579, November, 1974. Original article submitted April 30, 1974.

©1976 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

cess is carried out in neutral media. In the absence of sodium carbonate, the role of the base necessary for conversion of salt III to imine IV is played by the starting amine. As a result, ~50% of the amine is regenerated in the form of the hydrohalide. The absorption bands characteristic for NH and CO groups are absent in the IR spectra of Va-c and VII, whereas a number of bands due to vibrations of C=C and C=N bonds are present at 1580-1640 cm^{-1} .

Fusion of amine I with methyl chloroacetate at 125° for 10 min immediately gave oxo derivative VI, in the IR spectrum of which the stretching vibrations of the CO group appear at 1720 cm^{-1} .